

VOLODCHENKO, K.G.; BOMAS, O.V.; ISAKOV, L.I.; SMIRNOV, V.A.; KUNICHENKO, M.S.; LASHKOVA, Ye.A.; UVAROVA, M.A.; CHEVOTKINA, M.A.; NIKOLAYEV, P.S., glavnyy red.; SEREBRYAKOV, L.P., glavnyy red.; DERZHAVINA, N.G., red.; GUROVA, O.A., tekhn.red.; IVANOVA, A.G., tekhn.red.

[ENV unified production norms for operations in geological prospecting; mining operations] Edinye normy vyrabotki na geologorazvedochnye raboty (ENV); gornoproduktsionnye raboty. Moskva, Gos.nauchno-tekhn.izd-vo lit-ry po geol. i okhrane neдр, 1959. 123 p. (MIRA 13:6)

1. Russia (1923- U.S.S.R.) Ministerstvo geologii i okhrany neдр.
2. Otdel ekonomiki geologorazvedochnykh rabot Vsesoyuznogo nauchno-issledovatel'skogo instituta mineral'nogo syr'ya (VIMS) (for Volodchenko, Bomas, Isakov, Smirnov, Kunichenko, Lashkova, Uvarova, Chevotkina).

(Mining engineering--Standards)

MUKHAMETCVA, G.M., kand. med.nauk, otv. red.; GELLEN, L.I., kand.
med. nauk, red.; GIMADEYEV, M.M., red.; MIKHAYLETS, G.A.,
doktor med. nauk, red.; CHEVPETSOV, V.R., red.

[Industrial hygiene and health protection for the workers
of the petroleum and petrochemical industries] Gigiena
truda i okhrana zdorov'ia rabochikh v neftianoi i nefte-
khimicheskoi promyshlennosti. Ufa. Vol. 2. 1963. 547 p.
(MIRA 18:3)

1. Ufimskiy nauchno-issledovatel'skiy institut gigiyeny i
profzabolevaniy. 2. Direktor Ufimskogo nauchno-issledova-
tel'skogo instituta gigiyeny i profzabolevaniy (for Mukhame-
tova).

CHEVPILO, I.A. [Chevpylo, I.A.]

Effect of sulfur-containing amino acids on the synthesis of lipids.
Ukr.biokhim.zhur. 30 no.3:378-383 '58. (MIRA 13:3)

1. Institute of Biochemistry of the Academy of Sciences of the
Ukrainian S.S.R., Kiev.
(AMINO ACIDS) (LIPIDS) (SULFUR COMPOUNDS)

USSR/Diseases of Farm Animals. Diseases Caused by Helminths

R

Abs Jour : Ref Zhur - Biol., No 19, 1958, No 88291

Author : Chevchilo I.A.

Inst : Kiev Veterinary Institute

Title : The Problem of the Morphology of the Changes in the Pro-
ventriculus of Ducks Affected with Tetramerosis.

Orig Pub : Tr. Kiyevsk. vet. in-t, 1957, 13, 211-214

Abstract : As zones of compound glands which are situated within the thick stomach wall were histologically examined in proventriculi taken from 3 duckling carcasses, changes were established which are characteristic for diffuse catarrhal-desquamative processes. On the surface of the epithelium, these changes were characteristic of necrotic desquamative processes. Massive proliferation of connective tissue above superficial circulatory muscle layers were observed in 2 ducklings.

Card : 1/1

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CHEVPILO, I. A., Candidate of Biol Sci (diss) -- "The effect of sulfur-containing amino acids on ketogenesis and synthesis of lipids in the organism". Kiev, 1959. 13 pp (Min Agric Ukr SSR. Ukr Acad Agric Sci), 150 copies (KL, No 21, 1959, 114)

GULYY, M.F., akademik; FEDORCHENKO, Ye.Ya.; PECHENOVA, T.N.; MATUSEVICH, L.I.;
CHEVPILO, I.A.; PRONINA, Z.V.; ZHURAVSKIY, N.I.; MATSUEA, G.Kh.

Activation of amino acids with the formation of aminoacyl-
phosphates in animal tissues. Dokl. AN SSSR 166 no.1:227-230
Ja '66. (MIRA 19:1)

1. Institut biokhimii AN UkrSSR. 2. AN UkrSSR (for Gulyy).
Submitted July 2, 1965.

CHEVREVIDI, S.K.

Orchard grass *Dactylis glomerata* L. as raw material in the brush
manufacture. Isv. AN Us.SSR no.1:127-128 '53. (MIRA 11:3)
(Kirghisistan--Orchard grass) (Broom and brushes)

CHEVRENIDI, S.Kh; BRONOVITSKIY, V.Ye.

Some data on the study of Dzungarian iris. Trudy Inst.bot.AN Uz.SSR
no.3:201-207 '55. (MIRA 10:1)
(Kermine--Iris) (Brooms and brushes)

~~CHEVREUIL, S. M.~~ BRONOVITSKIY, V.Ye.

Roots of the orchard grass *Dactylis glomerata* L. as material for
manufacturing brushes. Trudy Inst.bot.AN Uz.SSR no.3:209-214 '55.-
(MIRA 10:1)

(Orchard grass) (Brooms and brushes)

CHEVRENIDI, S.Kh.

USSR/General Division - History. Classics. Personalities.

A-2

Abs Jour : Ref Zhur - Biologiya, No 1, 1957, 40.

Author : F.Kh. Dzhangurazov and S.Kh. ~~Chevrenidy~~.

Inst :

Title : The Scientific and Pedagogical Activity of V.P. Drobov
(on His 70th Birthday).

Orig Pub : Izv. AN Uz SSR, 1956, No 1, 109-112.

Abst : The 50th anniversary of the scientific, pedagogic, and public activity of Prof. Vasilii Petrovich Drobar (born in 1885), a great authority on plant life of the forests and sandy deserts of Central Asia, and one of the oldest Soviet botanists. He was the first to describe in detail the plant life of the Leno-Aldanskiy watershed; he studied the fertility of the sands in the Pribalkhash Area, and the flora of Yakutsk ASSR, of the Zeravshanskiy and Kirgizskiy Mountain chains and so forth. He studied the composition, distribution, and reserves of tanning

Card 1/2

CHEVRENIDI, S. KH.

USSR/Cultivated Plants - Medicinal. Essential Oils. Toxins.

M.

Abs Jour : Ref Zhur - Biol., No 4, 1958, 15878

Author : S. Kh. Chevrenidi

Inst :

Title : A New Efficient Source of Saponin from the Flora of Central Asia.
(Novyy effektivnyy saponinonos iz flory Sredney Azii).

Orig Pub : Izv. AN UzSSR, 1956, No 8, 95-97.

Abstract : Leontice ewersmannii Rge (I) is a perennial herbaceous plant of the barberry family which is widely distributed in all the rayons of Central Asia. A macroscopic description of all the plant's parts is given. The tubers of I up to 500 gr in weight contain the alkaloids leontamon, leontidin and leontin, ~30% starch, 45.9 - 72.5% water and up to 38.5% saponins which are not found in the parts of the plant above the surface.

Card 1/1

CHEVRENIDI, S.Kh.

Utilizing the antiseptic properties of the Hungarian Iris (*Iris songorica* Schrenk). Uzb. biol. zhur. no.2:67-70 '58. (MIRA 11:10)

1. Institut botaniki AN UzSSR.
(Iris) (Wood preservatives)

CHEVRENIDI, S.Kh.

Ampelopsis as a tannid plant. Izv.AN Uz.SSR no.6:109-110 '56.
(MIRA 14:5)
(Ampelopsis) (Tanning materials)

CHEVRENIDI, S.Kh.

New tannid plant. Izv.AN Uz.SSR no.7:89-90 '56. (MIRA 14:5)
(Tanning materials) (Epilobium)

CHEVRENIDI, S.Kh.

Some problems in the biology of *Acanthophyllum gypsophiloides* Rgl.
and its introduction into cultivation. Izv. AN Uz. SSR no. 12:55-
62 '56. (MIRA 14:5)

(*Acanthophyllum*)

CHEVRENIDI, S. Kh.

Geranium rectum Trautv as tanning plant. Biol. Glav. bot.
sada no.40:112-114 '61. (MIRA 14:10)

1. Institut botaniki AN UzSSR, Tashkent.
(Geraniums)
(Tanning)

CHEVRENIDI, S.Kh.

Basic results of the research on knotweed *Polygonum coriarium*
Grig. Trudy TashGU no.187:193-209 '61. (MIRA 15:3)

1. Institut botaniki AN UzSSR.
(Uzbekistan--Knotweed)

CHEVRENIDI, S.KH.

Allium aflatunense, a decorative plant of the future. Uzb.
biol. zhur. 6 no. 4: 30-32 '62. (MIRA 16:7)

1. Institut botaniki AN UzSSR.
(UZBEKISTAN---ALLIUM)

ZAKIROV, K.Z.; CHEVRENIDI, S.Kh.

Preservation and expedient use of the gifts of nature. Bot.
zhur. 47 no.6:838-843 Je '62. (MIRA 15:7)

1. Institut botaniki AN Uzbekskoy SSR, Tashkent.
(Uzbekistan—Botany, Economic)

CHEVRENIDI, S.Kh.

Polygonum coriarium Grig. and its cultivation in Uzbekistan.
Bot.zhur. 47 no.11:1641-1647 N '62. (MIRA 16:1)

1. Institut botaniki AN Uzbekskoy SSR, Tashkent.
(Uzbekistan—Polygonum)
(Uzbekistan—Plant introduction)

CHEVRENIDI, S. Kh.; MUSTAFAYEV, S.

Dye plants of Surkhan-Dar'ya Province of the Uzbek S.S.R. Uzb. biol.
zhur. 8 no.4:58-63 '64. (MIRA 18:7)

1. Institut botaniki AN UzSSR.

CHEVRENIDI, S.Kh.; PAUZNER, L.Ye.; MART'YANOV, A.N.

Where science joins production. Rast. res. 1 no.1:128-129 '65.
(MIRA 18:6)

1. Otdel rastitel'nogo syr'ya Instituta botaniki AN UzSSR,
Tashkent.

PAUZNER, L.Ye.; SOKOLOV, P.D.; CHEVRENIDI, S.Kh.

Expedition to Central Asia for the study on *Polygonum coriarium*.
Rast. res. 1 no.1:157-158 '65. (MIRA 18:6)

1. Botanicheskiy institut im. V.L. Komarova AN SSSR, Leningrad,
1 Institut botaniki AN UzSSR, Tashkent.

ZAKIROV, K.Z.; MOTKHIN, I.N.; CHEVRENIDI, S.Kh.; GRANITSKY, I.I.,
prof., otv. red.; KVIATKOVSKAYA, V.V., red.

[Soaproot of Turkestan; its biology and the methods of
introducing it into culture] Turkestanskii myl'nyi koren';
voprosy biologii i puti vvedeniia v kul'turu. Tashkent,
Izd-vo "Nauka" UzSSR, 1965. 107 p. (MIRA 18:10)

CHEVRENIDI, S.Kh.; PAUZNER, L.Ye.; SOKOLOV, P.D.

Joint expedition to natural places of *Polygonum corianum*. Uzb.
iol. zhur. 9 no.4:69-70 '65. (MIRA 18:10)

1. Institut botaniki AN UzSSR.

CHEVTAYEV, I.

Closer to people, closer to industry. Okhr. truda i sots. strakh.
no.1:25-29 JI '58. (MIRA 11:12)

1.Sekretar' Sverdlevskogo oblastnogo soveta profsoyuzov.
(Sverdlevsk Province--Industrial hygiene)

CHEVTAYEV, Ya., starshiy instruktor peredovykh metodov truda

Precast reinforced concrete roofs. Stroitel' no.2:13-14
F '59. (MIRA 12:5)
(Roofing, Concrete)

PIPKO, B.; CHEVICHALOV, A.

In a 310-meter longwall. Mast.ugl. 9 no.4:18-19 Ap '60.
(MIRA 13:11)
(Donets Basin--Coal mines and mining--Labor productivity)

CHEVYCHALOV, A.

Two cycles per day. Mast. ugl. 9 no. 9:4-5 S'60. (MIRA 13:10)
(Donets Basin--Coal mines and mining--Labor productivity)

hh499
8/181/63/005/001/014/064
B102/B186

5900
AUTHORS: Cubanov, A. I., and Chevychelov, A. D.

TITLE: Theoretical estimates of the chain rupture energy in solid polymers

PERIODICAL: Fizika tverdogo tela, v. 5, no. 1, 1963, 91-95

TEXT: The fact that the rupture energy D is different in different polymers even if it is always C-C bonds that are ruptured (except polyamides wherein C-N is ruptured) shows that three factors influence D . These factors are studied here. (1) If the exchange integrals (A) and the repulsion of neighboring carbon ions ($\Delta Q = (Z' - 1)^2 / R$) are taken into account, then the change in the bond energy on the transition from a nonpolar polymer (e.g. polyethylene) to a polar one is $\Delta D = -\Delta A - \Delta Q$; for teflon $\Delta D = 10$ kcal/mole and $D = U_{C-C} + \Delta D = 69.6$ kcal/mole. R is the C-C distance, for teflon it is 2.92 Å, Z' is the effective nuclear charge. (2) Co-reactions may occur when a bond is ruptured, e.g., in polychlorovinyl where HCl arises on its thermal destruction. This reaction energy has, of course, to be taken into account when D is calculated. (3)

Card 1/2

The synthesis of *n*-chloropropionic acid by the condensation of phenacetone with ethylene. A. L. Kletnanski and K. E. Chivers. *Chem. Comp. rend. acad. sci. U. S. S. S.*, 2, 43-44 (in German 44-7) (1935). - C_2H_4 was dissolved in CCl_4 below 0° and $COCl_2$ and 0.5 mol. $AlCl_3$ gradually added. The condensation products, conig. $CH_2ClCHCl_2$, $COCl$, and the $AlCl_3$ sep'd. as a heavy dark layer. The CCl_4 was dist'd. off and anhyd. MeOH was added below 0°. This formed 80-90% $CH_2ClCHCl_2CO_2Me$, which after distn. is now still contained traces of 8 comp'd, from which it could not be sep'd., even over Na. Distn. at atm. pressure split out HCl to form $H_2C=CHCO_2Me$, which was treated with $SOCl_2$ and $PhNH_2$ to form *acrylamide*, m. 104°. Various unsuccessful attempts to condense these comp'ds. by other methods are mentioned, as is work by early men in the field. J. R. Milbery

J. E. Milbrer

The reaction of the combination of organic acids with monovinylacetylene to form butadiene esters. A. I. Kiselevskii and K. K. Chervinskaya. *Soviet. Khimich. No. 4, 5-13 (1955)*.—Monovinylacetylene formate was prepd. by admitting into a flask with agitator and reflux, and cooling, 4% of HgO , H_2O , BF_3 and HCO_2H , monovinylacetylene (I) from a dropping funnel. The yield of the ester (II) exceeded 64% after 4 hrs. Kapa. carried out with $ZnSO_4$, $CdSO_4$, $ZnCl_2$, ZnO , CuO and $Cu_2(CO_3OH)_2$, as catalysts were unsatisfactory. II had the formula $C_4H_6O_2$. Various tests of the II are reported. The reaction carried out with $AcOH$ and I under similar conditions yielded only 21% of the corresponding ester (acetoxypropene), of the formula $C_4H_6O_2$. According to various tests the product of the reaction of HCO_2H with I is 2-butenyl-1,3-butadiene (III) and that of $AcOH$ with monovinylacetylene is 2-acetoxy-1,3-butadiene (IV). Both III and IV form rubber-like polymers, the polymerization velocity of III exceeding that of IV. The process is described, and 16 references are appended. A. A. Bochtlingk

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Synthesis of *p*-chloropropionic and acrylic acids by oxidation of phosphono and allylphos. A. I. Kletanskii and K. K. Chukhlovskii. *J. Gen. Chem.* (U.S.S.R.) 5, 815-82 (1934).—A crit. discussion of the literature with more than 23 references is given. The neg. results obtained by Vorobavskii and Iurgenyevskaya (1: A. 28, 20831) in the formation of $\text{ClCH}_2\text{CH}_2\text{COCl}$ (I) by the condensation of COCl_2 with C_3H_6 in PhMe and CCl_4 with AlCl_3 as catalyst are confirmed. Similar neg. results were obtained in PhCl solvent. Various conditions of Friedel-Crafts condensation of COCl_2 and C_3H_6 , and isolation of I attempted are described in detail. The condensation of COCl_2 with C_3H_6 with the formation of I is considered as basically possible. The optimum conditions are at the concn. of 0.5 g./mol. of AlCl_3 in CCl_4 with cooling and stirring. The product of the reaction is, apparently, I, which cannot be isolated as such by the decompos. of the complex. The neg. of the condensation products may be more conveniently effected in the form of the esters of $\text{ClCH}_2\text{CH}_2\text{COOH}$ (II) by decompos. the AlCl_3 complex with alc. a/c. In addn. to II a considerable formation of the products of polymerized C_3H_6 takes place. No improvement is effected by carrying out the reaction under pressure or by substituting ZnCl_2 for AlCl_3 . Chas. Blanc

ASSN. SLA METALLURGICAL LITERATURE CLASSIFICATION

FROM: SYNGINESE

SECOND MAP ONLY CRI

RELATION

FROM: BUNYON

RELATION ON ONE ALL

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The synthesis of the derivatives from 2,4-dichloro-2-butene. The use of by-products from the synthesis of chloroprene. A. I. Klebanik and K. K. Chervukhalova. *Soviet. Kauchuk* 1933, No. 6, 10-21. (1) 2,4-dichloro-2-butene (I) was treated with a 15% excess of 24-30% alc. KOH, by stirring for 4 hrs. on a water bath (60-70°). The product of reaction was sepd. from salts and fractionated *in vacuo* and was collected at 102-104° at 40 mm.; it had the following const.: d_{40}^{20} 0.7920, n_D^{20} 1.4392, mol. refraction 30.05, Cl 20.07%, C 61.04%, H 8.01%, formula C_4H_5OCl (Rt ether of 2-chloro-2-buten-4-ol); it did not polymerize. (2) I was treated with 15% excess of 15-20% Na_2CO_3 , by stirring for 2 hrs. on a boiling water bath. The product of reaction is a chloro alc. (chlorobutenol) (II); it was distd. at 90-110° (yield 70-80%), and $b.p.$ 92°, d_{40}^{20} 1.101, n_D^{20} 1.4032, mol. refraction 26.64, Cl 32.8%, C 45.33%, H 9.47%, O 15.40%, formula C_4H_5OCl ; it did not polymerize. (3) To establish the alc. group in II a xanthate was obtained by adding the theoretical amt. of KOH to the II with stirring for 30 mins. and heating on a water bath. KOH was in a suspension; with the addn. of a 20% excess of Cl₂, a crystal.

ppt. of xanthate was formed; the latter was dried in air or at const. temp. of 30-40°. The reaction proceeded as follows: $CH_2ClCHClCH_2OH + KOH + CS_2 \rightarrow KSCOCH_2CHClCH_2Me + H_2O$. The resulting xanthate was sol. in water and alc.; it contained Cl 17.6%, S 31.29%, at 142° it changed from yellow to a darker color; it can be used for flotation (equals the K Rt xanthate). (4) The water soln. of KOH (15% in excess) was treated with II, heated on a water bath, and I was gradually added with stirring. The reaction was completed in 4 hrs. The product of reaction was washed with water, dried over $CaCl_2$ and distd. *in vacuo* (yield 80-90%). The const. of the resulting dichlorodibutenyl ether were $b.p.$ 142°, Cl 30.9%, d_{40}^{20} 1.171, n_D^{20} 1.4890, mol. refraction 30.11, formula $C_4H_2Cl_2O$. The product did not polymerize. (5) I was passed through a china pipe over silica gel at 245-255°, over Glukhovskaya clay at 240-260° and 250-300° and over molten KOH at 265-450°. In all cases chloroprene was obtained (yield 17-24%), with the following const.: $b.p.$ 20°, d_{40}^{20} 0.9578, n_D^{20} 1.4570.

A. Pestoff

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PROCESSES AND PROPERTIES INDEX

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The synthesis of β -chloropropionic and acrylic acids by condensation of phosgene with ethylene. A. I. Kleban-
skii and K. K. Chevychalova. *Trans. State Inst. Applied Chem.* 1937, No. 3, 46-52; *Khim. Referat. Zhur.* 1938, No. 6, 92.—Condensation of phosgene with ethylene was performed in the presence of catalysts ($ZnCl_2$, $AlCl_3$, etc.). The effects of the solvent and the temp. of the reaction on the yield of the final products were investigated. Optimum results were obtained from $AlCl_3$ with a gradual addition of the catalyst and with mixing. The optimum concn. of $AlCl_3$ was 0.5 mol./mol. of each of the reaction substances. CS_2 was used as the solvent. The use of toluene, $PhCl$ and CCl_4 for this purpose was rejected owing to their instabilities under the conditions of the expts. (Friedel-Crafts reaction). K. and C. consider that the $ClCH_2CH_2COCl$ was the final product of the reaction, although it was not possible to sep. this product in the pure state, but approx. 30% (based on ethylene) of the $ClCH_2CH_2CO_2Et$ was obtained from the decompn. of the prod-
ucts of the reaction with abs. alc. A considerable amt. of the high-boiling unsatd. hydrocarbons was obtained to-
gether with $ClCH_2CH_2CO_2H$. An increase of the pressure did not increase appreciably the yield of the acid. $ZnCl_2$, silica gel and activated charcoal were not suitable for this reaction.

W. R. Hunt

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ASAC ELEMENTS

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The synthesis of β -chloropropionic and acrylic acids by condensation of phosgene with ethylene. A. I. Kleban-
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W. R. Hunt

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ASAC ELEMENTS

PROCESSES AND PROPERTIES INDEX

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Carboxy derivatives of butadiene. Their polymerization and the properties of the polymers. A. L. Kabanikh and K. K. Chyrychakova. *J. Gen. Chem. (U.S.S.R.)* 16, 1101-14 (1946) (in Russian).—The prepn. of acetoxybutadiene (I) was studied over $\text{HgO} \cdot \text{BF}_3 \cdot \text{Et}_2\text{O}$ catalyst, 80% being the best yield. A cyclic ether ketone was isolated which was the dimerization product of I and formylacetylene (II). Polymerization rates for I and II derive were detd. at 20, 40, and 80°; II polymerizes twice as fast as I. Doubling the temp. increases the polymerization rate by a factor of 2. Ozonolysis of the polymers gave succinic acid and acids of the corresponding esters, thus showing that the polymers have the structure type, 1-4-1-4 or 1-4-1-1. Vinylacetylene (III) could not be condensed with HgO in the presence of the above catalyst; condensation with PhOH , however, took place but the product could not be isolated in a pure state nor characterized. Hydrolysis of I or II with alc. alkali gave polybutadienyl alc. (IV), which was sol. in water, EtOH , CHCl_3 , AcOH , and pyridine. A 25-100% excess of III was passed into AcOH conig. 5-10% $\text{HgO} \cdot \text{BF}_3 \cdot \text{Et}_2\text{O}$ with 5% Ac_2O being used as an activator; the reaction was carried out below 14°, the final temp. being 20° after a 2-6 hr. run; the product was isolated by extn. with Et_2O , followed by washing with NaCl soln., NaHCO_3 , and water, and drying; 1, b.p. 83-4°, n_D^{20} 1.4418, d_4^{20} 0.9046, gives a *maleic anhydride* condensate, m. 169-71°, which appears to be 4-acetoxytetrahydrophthalic acid. Use of CHCl_3 or petr. ether for the initial reaction gave poor yields, while Et_2O inhibited the reaction completely. The residue after iso-

lation of I gave, after fractionation, a product, b. 135-6°, n_D^{20} 1.4476, d_4^{20} 1.0087, which corresponds to a condensate of 2 moles. I deriv., with hydrolysis of 1 acidic residue on the double bond and isomerization of the resulting alc. into a ketone; the product gives a *semicarbazone*, m. 188-9° (from water), while hydrolysis by 0.5 N HCl on a steam bath gave a *diketone*, which was isolated as a *semicarbazone*, m. 210-17°; thus the product may be either a 1,3- or 1,4-isomer of tetrahydrooctadecapentaphenone. A similar product, b. 114°, n_D^{20} 1.4700, d_4^{20} 1.1011, was obtained from distn. residues in the prepn. of II. Polymers of I or II derive. were hydrolyzed by theoretical amts. of 7-10% alc. NaOH , using PhMe for preliminary swelling of the polymers; hydrolysis was conducted at room temp. until the material formed a uniform soln., preferably in an inert atm. and in the presence of an antioxidant; the mixt. was then neutralized with HCl , the solvents were removed *in vacuo*, and the residue taken up in water-insol. dialyzed; IV was occasionally obtained as a water-insol. yellow mass, in which case it was purified by multiple soln. in EtOH and pptn. with water; it readily formed a *polybenzoyl deriv.* by treatment with excess BaCl_2 in pyridine with heating 12 hrs. on a steam bath; after soln. in CHCl_3 and pptn. with Et_2O it formed a brown powder, sol. in CHCl_3 and PhOMe , insol. in EtOH , Et_2O , petr. ether. G. M. Kosolapoff

Structure of polychloroprene as determined by osmoly-
sis. M. A. Kabanikh and K. Chesychalova. J.
Gen. Chem. (U.S.S.R.) 17, 941-95 (1967); cf. C.A. 61,
60217. Five chloroprene (I) polymers were prep'd. as
follows: (1) I was polymerized in the presence of a regu-
lator (not described) at 40°. Polymerization was stopped
at a yield of 76%. The polymer was -d. in CHCl₃.
(2) Emulsion polymerization of I was carried out in the
presence of antioxidants. The polymer yield was limited
to 40% so that CHCl₃-sol. polymer was obtained. (3) A
latex was made by Carothers' method (C.A. 56, 78),
which involved polymerization of an emulsion of I in a 2%
aq. soln. of Na idrate. (4) I was placed in a sealed tube
and heated at 80° for 1 hr. The polymer yield was 30%.
(5) Liquid I polymer was made by an industrial poly-
merization process involving the use of regulators, but not
otherwise described. Each of the 5 polymers was sub-
jected to ozonolysis by dissolving it in CHCl₃ and bubbling
O contg. O₃ into the soln. until the mixt. lost its capacity
to react with Br. The ozonides formed were decompd.
with aq. H₂O, and the products analyzed. In every case
87% or more of the hydrocarbon skeleton could be identi-
fied as succinic acid (II), HCO₂H, or CO₂ in the final mixt.
(4) these 3 products, II represented approx. 95% in every
case. This result indicated that all 5 polymers were 1,4-
addn. polymers of I. The percentage of the hydrocarbon
skeleton identified was lowest in the polymers made by
methods (3) and (5), i.e., 85-88%, compared to 95-
99.6% for the other 3 polymers. The product from
method (3) was believed to contain an appreciable insol.
polymer and also emulsifying agents and the like that did
not respond in the normal way to ozonolysis. The liquid
polymer was thought to have given low results because of
the presence of unidentified reaction products between the
polymer and the regulator. Expts. were also carried out
in which the ozonides were decompd. with HCl in abs. alc.

with or without Et orthoformate. The products were not
completely identified but contained Et succinate, the semi-
acetal of II, and a fraction with a compn. corresponding in
analysis and mol. wt. to impure C₁₂H₁₄O₄CH₂CH₂CHO
dimer, possibly representing a mixt. of the monomer and
trimer of this compd. No deriva. of II dialdehyde,
(CH₂CHO)₂, were identified, although an intensive search
was carried out for the diacetal of this compd. The use
of the reactivity with Br as an index for detg. the com-
pletion of ozonolysis was investigated critically. The
consumption of Br was approx. const. up to the end of
ozonolysis. Thereafter the consumption decreased
sharply. It was observed that Br tended to react with the
ozonolysis mixt. by substitution, since HBr was evolved
in amts. as large as 33% of the total Br.

H. K. Livingston

13

29

Structure of Chloroprene Polymers as Determined by
Osmolysis. XI. A. Klebanovskii and K. Chepychalova,
Rubber Chemistry and Technology, v. 21, July 1948,
p. 605-620. Translated from Zhurnal Obshchei
Khimii (Journal of General Chemistry), v. 17, no.
5, 1947, p. 941-966.
Previously abstracted from original source.

ASTM-SLA METALLURGICAL LITERATURE CLASSIFICATION

14000 15000 16000 17000 18000 19000 20000 21000 22000 23000 24000 25000 26000 27000 28000 29000 30000 31000 32000 33000 34000 35000 36000 37000 38000 39000 40000 41000 42000 43000 44000 45000 46000 47000 48000 49000 50000 51000 52000 53000 54000 55000 56000 57000 58000 59000 60000 61000 62000 63000 64000 65000 66000 67000 68000 69000 70000 71000 72000 73000 74000 75000 76000 77000 78000 79000 80000 81000 82000 83000 84000 85000 86000 87000 88000 89000 90000 91000 92000 93000 94000 95000 96000 97000 98000 99000

branching takes place, with the formation of three-dimensional
linear polymers, which have linear properties similar to those of
gutta-percha. No precise details for the formation of the polymers
are given. The polymer of 2 : 3-dichlorobuta-1 : 3-diene is
removed by suspending it in CHCl_3 and passing O_2 into the suspen-
sion slowly for 48 hr. until the solution no longer darkens. It.
The residue obtained was decomposed by heating with H_2O
slowly. CH_2O (18%), HCO_2H (8-12%), and succinic acid (42-78%)
being obtained. It is assumed that there is also decomposition of
the C chain to CO_2 and HCl which are not recovered. The polymer
of 1 : 2 : 3-trichlorobuta-1 : 3-diene is oxidized in CHCl_3 (48 hr.),
and the residue is decomposed with H_2O , (18%) H_2O on the water-
bath for 4 hr. No aldehydes are observed. 81% of organic acids
are obtained, m.p. 266-267°, corresponding to fumaric acid, formed
by the decarboxylation of 1-chlorosuccinic acid, which is to be expected
if the chief polymer unit is of the structure $\text{>CCl-CH}_2\text{-CHCl-CCl<}$.
C. A. Price.

5 (3)

SOV/79-29-3-16/51

AUTHORS:

Garmonov, I. V., Klebanskiy, A. L., Chevychalova, K. K.

TITLE:

Preparation of Div inyl by the Catalytic Hydrogenation of Vinyl Acetylene (kataliticheskoye gidrirovaniye vinilatsetilena s tsel'yu polucheniya divinila). I. General Kinetic Rules of the Selective Hydrogenation of Vinyl Acetylene in Solution (I. Obshchiye kineticheskiye zakonomernosti izbiratel'nogo gidrirovaniya vinilatsetilena v rastvore)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 3, pp 824-830 (USSR)

ABSTRACT:

The authors thoroughly investigated the catalytic hydrogenation of vinyl acetylene in order to increase the selectivity of this reaction and to find the technological basis of this process for its industrial utilization. On investigating the composition of the hydrogenation products of vinyl acetylene in the solution with various catalysts it could be found that the palladium catalyst produces the highest selectivity on the hydrogenation. This capability is illustrated by the following graduation order: palladium->iron skeleton->nickel skeleton->platinum black catalyst, which is in contrast with references 3 and 4. The hydrogenation with the palladium catalyst in the

Card 1/3

SOV/79-29-3-16/61

~~Preparation of Divinyl by the Catalytic Hydrogenation of Vinyl Acetylene~~
~~1. General Kinetic Rules of the Selective Hydrogenation of Vinyl~~
acetylene in Solution

vapor phase at 130° was not possible owing to side reactions. Therefore the hydrogenation of vinyl acetylene was carried out on the palladium catalyst with finely powdered silica gel as carrier. It takes place in the first step of the process on vinyl acetylene and on hydrogen. In the kinetic range of hydrogenation the reaction rate is directly proportional to the quantity of the catalyst and does not depend on the intensity of stirring of the solution. In the diffusion range on the hydrogen the reaction rate increases proportionally to the increasing intensity of stirring of the solution and does not depend on the quantity of the catalyst. In both hydrogenation ranges the reaction rate increases proportionally to the increasing partial pressure of the hydrogen. In order to find out those conditions which produce the highest selectivity in the process and to facilitate the separation of the principal product of the reaction, the divinyl, (butadiene-1,3) in a pure state, the composition of the reaction products obtained at different intensity of hydrogenation was determined. In the initial stage of the process, up to a hydrogenation intensity

Card 2/3

SOV/79-29-3-16/61

Preparation of Divinyl by the Catalytic Hydrogenation of Vinyl Acetylene
1. General Kinetic Rules of the Selective Hydrogenation of Vinyl
Acetylene in Solution

~30% (calculated with respect to the acetylene bond) the
affiliation of the hydrogen was found to take place mainly to
the triple bond. On further hydrogenation in addition to this
affiliation a hydrogenation of the divinyl being formed takes
place wherein the reaction products represent a very complex
mixture of hydrocarbons which are difficult to separate. By
hydrogenation of the mixture of vinyl acetylene and divinyl
the above mentioned reaction character was confirmed. On a low
intensity of hydrogenation (up to 30%) practically only di-
vinyl is obtained. There are 4 figures, 2 tables, and 9 ref-
erences, 5 of which are Soviet.

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo
kauchuka (All-Union Scientific Research Institute of Synthetic
Rubber)

SUBMITTED: February 3, 1958

Card 3/3

5 (3)
AUTHORS:

SOV/79-29-3-17/61
Garmonov, I. V., Klebanskiy, A. L., Chevychalova, K. K.

TITLE:

Preparation of Divinyl by the Catalytic Hydrogenation of Vinyl Acetylene. I. Influence of Various Factors Upon Rate and Selective Behavior of Hydrogenation of Vinyl Acetylene (I. Vliyaniye razlichnykh faktorov na skorost'i izbiratel'nost' gidrirovaniya vinil-atsetilena s tsel'yu polucheniya divinila). I. Influence Exerted by Various Factors Upon Rate and Selective Behavior of Hydrogenation of Vinyl Acetylene (I. Vliyaniye razlichnykh faktorov na skorost'i izbiratel'nost' gidrirovaniya vinil-atsetilena)

PERIODICAL:

Zhurnal obshchey khimii, 1959, Vol 29, Nr 3, pp 830-836 (USSR)

ABSTRACT:

In the present paper the authors present data on the influence exerted by the nature of the carrier and solvent, the reaction temperature and the intensity of the mixing of the solution upon the rate and the selective behavior of the hydrogenation of vinyl acetylene. In order to investigate the influence of the conditions at the preparation of the catalyst and that of the character of the carrier upon the rate and the selective behavior of this hydrogenation, experiments with palladium on silica gel, with barium sulfate and with polyvinyl alcohol were carried out. It was found that a modification of the preparation conditions of the catalyst and of the nature of the carrier

Card 1/3

SOV/79-29-3-17/61

Preparation of Divinyl by the Catalytic Hydrogenation of Vinyl Acetylene
I. Influence Exerted by Various Factors Upon Rate and Selective
Behavior of Hydrogenation of Vinyl Acetylene

influences the reaction rate, but that the selective behavior of hydrogenation is not markedly influenced by these modifications. The results obtained theoretically completely correspond with the publications available in this field (Ref 2). From among all factors investigated the intensity of the mixing of the solution and the percentage of the quantity of the medium exert a noticeable influence upon the selectivity of the reaction. At a low intensity the reaction did not proceed selectively. In the alcohol solution with the $\text{pH} > 7$ the reaction proceeded more rapidly but with less selection than in acid and neutral medium. The determined hydrogenation character of the dissolved vinyl acetylene in the presence of the palladium catalyst as well as the determined dependence of the selective behavior of the process on the intensity of the mixing of the solution completely agree with the absorption theory concerning the catalytic hydrogenation (Ref 4). There are 1 figure, 5 tables, and 5 Soviet references.

Card 2/3

SOV/79-29-3-17/61

Preparation of Divinyl by the Catalytic Hydrogenation of Vinyl Acetylene
I. Influence Exerted by Various Factors Upon Rate and Selective
Behavior of Hydrogenation of Vinyl Acetylene

ASSOCIATION: Vsesoyuznyy nauchno-issledovatel'skiy institut sinteticheskogo
kauchuka (All-Union Scientific Research Institute of Synthetic
Rubber)

SUBMITTED: February 3, 1958

Card 3/3

39845

S/190/62/004/008/003/016
B117/B144

5.3832

AUTHORS: Klebanskiy, A. L., Chevychalova, K. K., Yefremova, Ye. M.

TITLE: Formation conditions and structure of dimers obtained by the bulk polymerization of chloroprene

PERIODICAL: Vysokomolekulyarnyye soyedineniya, v. 4, no. 8, 1962, 1145-1150

TEXT: The polymerization of chloroprene in the presence of peroxides or tetraalkyl Thiuram disulfide with sulfur at 50 - 55°C, and the effect of individual factors on the formation of dimers, were studied. The formation rate and the dimer-to-polymer ratio increase when the polymerization temperature rises. Dimers are the main products obtained in the presence of inhibitors. Increasing the degree of polymerization to >70% brings down the relative amount of the resulting dimers to 4 - 5%. The high-molecular polymerization of chloroprene in the presence of tetramethyl thiuram disulfide yielded dimers, of which 65% have the structure of 1,5-dichlorocycloocta-1,5-diene, whilst 12% were 4-(1-chlorovinyl)-1-chlorocyclohexene. The structure was determined by oxidation, ozonolysis, and hydrogenation at 20°C in the presence of platinum black, yielding a

Card 1/2

1 60207-65 ENT(m)/EPP(6)/EWP(3)/T PC-4/Pr-4 GS/JAJ/RM
ACCESSION NO. ATSO19606

1. 60207-65 ENT(m)/EPP(6)/EWP(3)/T PC-4/Pr-4 GS/JAJ/RM

2. 60207-65 ENT(m)/EPP(6)/EWP(3)/T PC-4/Pr-4 GS/JAJ/RM

3. 60207-65 ENT(m)/EPP(6)/EWP(3)/T PC-4/Pr-4 GS/JAJ/RM

TOPIC TAGS: contaminant, triisobutyl aluminum, isoprene, polymer, rubber, Ziegler catalyst

ABSTRACT: The effect of contaminants present in a reaction mixture of triisobutyl aluminum, isoprene, and Ziegler catalyst was studied. The contaminants studied were water, ethyl alcohol, and ethyl acetate. The concentration of the catalyst was 10-20 vol % based on reaction mixture and the concentration of isoprene was 10-20 vol % based on isoprene. After 1 hour of reaction, the reaction mixture was poured with ethyl alcohol. The effect of the contaminants on the reaction was studied.

Cord 1/2

67304

24.7700

~~24(3), 9(3)~~

AUTHOR:

Chevychelov, A. D.

SOV/181-1-8-9/32

TITLE:

Volt-ampere Characteristic of an Electron - Hole Junction²¹
Under Consideration of Generation and Recombination of Current
Carriers in a Space-charge Layer

PERIODICAL:

Fizika tverdogo tela, 1959, Vol 1, Nr 8, pp 1205 - 1212 (USSR)

ABSTRACT:

For the volt-ampere characteristic the expression
 $I = I_0(e^{\tilde{u}} - 1)$ holds according to the present theory of
electron-hole junction. From this formula follows the exist-
ence of a saturation current in the backward direction for
the case that $\tilde{u} \rightarrow -\infty$. In this theory recombination of elec-
trons and holes in the n- and in the p-domain is considered.
The authors point out several shortcomings in the theory of
Sah, Noyce, and Shockley (Ref 4). In the first part the equa-
tions for electron-hole junction with symmetric potential are
derived considering production and recombination processes in
the transition region. The one-dimensional diffusion-recombi-
nation equations for electrons and holes in the steady case

Card 1/3

67304

Volt-ampere Characteristic of an Electron - Hole SOV/181-1-8-9/32
Junction Under Consideration of Generation and Recombination of Current
Carriers in a Space-charge Layer

read $dj_n/dx + U = 0$, $dj_p/dx + U = 0$, U denoting the variation in electron- or hole number per unit volume and unit time. After consideration of the respective boundary conditions one obtains equations for the determination of the integration constants. The authors then solve the problem for a concrete potential. Impurity concentration varies linearly in the diffusion n-p-junctions. For n-p-junctions with such a distribution the potential may be well approximated by a cubic parabola. In the investigation of generation-recombination processes only monomolecular recombination through the middle of the recombination centers is considered. Their concentration remains constant along the n-p-junction and they form a level in the forbidden zone. In the second paragraph of the present paper the voltampere characteristic for forward and backward current direction is derived.

$$U(x) = \bar{U} = (1/2d) \int_{-d}^d U(x) dx = \lambda U(0) \text{ serves as zeroth approxim-}$$

Card 2/3

67304

Volt-ampere Characteristic of an Electron - Hole Junction Under Consideration of Generation and Recombination of Current Carriers in a Space-charge Layer SOV/181-1-8-9/32

ation. \bar{U} denotes the mean value of $U(x)$ within the transition region and λ is a slowly variable function of u . A numerical example is computed for silicon. When the ratio d/L (d denoting the half width of the transient region and L the diffusion length) is of the order one, then the carrier concentrations cannot vary much because of the recombination in the transition layer and thus the concentration values resulting from the usual theory may be employed. Voltampere characteristic depends only slightly on potential shape in the transition region and the same result is obtained for extremely steep and flat potentials. The author thanks A. I. Gubanov for valuable remarks during the work and for his help in printing the manuscript. There are 1 figure, 2 tables, and 5 references, 2 of which are Soviet. Leningradskiy fiziko-tekhnicheskii institut AN SSSR (Leningrad Physical and Technical Institute of the AS USSR) August 5, 1958

ASSOCIATION:

SUBMITTED:

Card 3/3

Chevychelov, A. D.

82531

24.7100
24.6200

S/181/60/002/007/003/042
B006/B070

AUTHORS: Gubanov, A. I., Chevychelov, A. D.

TITLE: Calculation of the Energy Spectrum of Strongly Anisotropic Crystals 71

PERIODICAL: Fizika tverdogo tela, 1960, Vol. 2, No. 7, pp. 1379-1389

TEXT: The purpose of the present work is to make a quantum-mechanical calculation of the energy spectrum of the electrons for two different models of an anisotropic crystal. In their theory of galvanomagnetic phenomena in metals I. M. Lifshits, M. Ye. Azbel', and M. I. Kaganov (Ref. 1) have assumed the existence of open isoenergetic surfaces in the k-space. The authors of the present paper have succeeded in establishing similar surfaces theoretically by investigating two models of a strongly anisotropic crystal: layer model and chain model. In a crystal with layer-model structure, for example, zinc, such planes appear since the valence forces are mainly acting between the atoms that lie in these planes; the binding between the planes is considerably weaker, and the

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82531

Calculation of the Energy Spectrum of
Strongly Anisotropic Crystals

S/181/60/002/007/003/042
B006/B070

interatomic distance in the directions perpendicular to the layers are correspondingly large. In the direction of this crystallographic axis (z-direction), the motion of the electrons (in the single-electron crystal model) is assumed to be almost free, and in the other two directions strongly inhibited. Analogous assumptions are made for the chain model: A number of crystals have chain structure, that is, they have long spiral chains coiled round the hexagonal axis (selenium, tellurium), and the valence forces act along these chains; the binding between these chains is relatively weak, and the interatomic distances are large. The expressions for the potential of the electron is the same in both cases, with the difference only that in the first case the component $V(\vec{r})$ occurs, which is a two-dimensional periodic function, and in the second case the function $V(z)$ occurs which is periodic only in the z-direction. For both models, the spectrum is investigated by starting from the Schrödinger equations for these potentials. Later, the authors consider the shapes of the isoenergetic surfaces, first for the hexagonal lattice both for the layer and the chain models (Fig. 1). Fig. 2 shows the functions $\epsilon_1(\vec{q})$ and $\epsilon_2(p)$; $\vec{q} = (k_x, k_y)$ being the two-

Card 2/4

8 2531

Calculation of the Energy Spectrum of
Strongly Anisotropic Crystals

S/181/60/002/007/003/042
B006/B070

dimensional k-vector and p its z-component. $\epsilon_1(\vec{q})$ has a band shape, and for the first, third, and generally for odd bands $\epsilon_1(\vec{q})$ has a minimum in $\vec{q} = 0$ and a maximum in $\vec{q} = \vec{Q}$ (\vec{Q} is a vector that lies in the boundary for the first Brillouin zone). For even bands $\epsilon_1(\vec{q})$ has a minimum in $\vec{q} = \vec{Q}$ and a maximum in $\vec{q} = 0$. Later, the isoenergetic curves for the planes formed by p and \vec{Q} are investigated. Fig. 3 shows some possible odd bands. It is seen that the topology of the isoenergetic surfaces for the two models are opposite to each other. For sufficiently large depths of potential both models have open surfaces of only one kind: corrugated cylinders in the case of layer structure, and corrugated planes in the case of chain structure. The theoretical results agree with the experimental ones. It is further shown that for lattices with the same crystallographic symmetry, but different chemical binding characteristics different laws hold for the dispersion of electrons. There are 3 figures and 4 Soviet references.

ASSOCIATION: Fiziko-tehnicheskiy institut AN SSSR Leningrad
(Institute of Physics and Technology of the AS USSR,
Leningrad)

Card 3/4

Calculation of the Energy Spectrum of
Strongly Anisotropic Crystals

SUBMITTED: November 5, 1959

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B006/B070

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Card 4/4

5,4100

36872

S/181/62/004/004/013/042

B104/B108

AUTHORS: Cubanov, A. I., and Chevychelov, A. D.

TITLE: Theory of the breaking strength of solid polymers

PERIODICAL: Fizika tverdogo tela, v. 4, no. 4, 1962, 928 - 933

TEXT: This is a critical comment of P. Bueche's theory (J. Appl. Phys., 28, 784, 1957). The theoretical strength of a polymer is calculated on the assumption that the potential energy of interaction between neighboring atoms of polymer chains can be described by a Morse function

$$U(r) = D \left[\exp(-2(r-R)/a) - 2\exp(-(r-R)/a) \right].$$

D is the maximum depth of the potential well; a characterizes the curvature of U(r) near its minimum, and R is the equilibrium interatomic distance. For the time until a sample breaks under a given load, the following relation is obtained:

$$\ln(\tau/\tau_0) = D/kT - \ln \left\{ \exp \left[(a\sigma e/kT) (1 + \ln(2DN/a\sigma)) \right] - 1 \right\}, \text{ where } \tau_0 = 1/\omega,$$

Card 1/2

Theory of the breaking strength ...

S/181/62/004/004/013/042
B104/B108

N is the total number of chains passing through unit cross section. In the case of polyvinyl chloride, polypropylene, and polyethylene, the calculated strength is considerably greater than the experimental one. Caprone is an exception. These results diverge from experimental data less than Bueche's results. Explanation: (1) Since polymer chains have finite dimensions, the effective value of N is influenced thereby; (2) irregularity was considered through the factor $1/3$ in the calculation. This factor may be lower in an exact calculation. (3) The sample displays inhomogeneities. A fluctuation mechanism is assumed to be the principal cause of polymer destruction. In these calculations, intermolecular forces were assumed to be small. S. N. Zhurkov, Corresponding Member AS USSR, is thanked for having suggested the subject and for discussions. There are 3 figures and 1 table. f

ASSOCIATION: Fiziko-tehnicheskiy institut im. A. F. Ioffe AN SSSR
Leningrad (Physicotechnical Institute imeni A. F. Ioffe,
AS USSR, Leningrad)

SUBMITTED: November 23, 1961
Card 2/2

CHEVYCHELOV, A.D., GUBANOV, A.I.

Precise formulation of the kinetic theory of polymer strength.
Bond and cohesive energy in polymers.

Report presented at the 13th Conference on High-molecular compounds
Moscow, 8-11 Oct 62

GUBANOV, A. I.; CHEVYCHELOV, A. D.

Theoretical estimates of the bond breaking energy in solid
polymers. Fiz. tver. tela 5 no.1:91-95 Ja '63.
(MIRA 16:1)

1. Fiziko-tehnicheskiy institut imeni A. F. Ioffe AN SSSR,
Leningrad.

(Chemical bonds) (Polymers)

L 13036-63 EPR/EWP(j)/EPF(c)/EWT(m)/BDS/ES(s)-2 AFFTC/ASD/ESD
 Ps-4/Pc-4/Pr-4/Pt-4 RM/WW
 ACCESSION NR: AP3000619

S/0181/63/005/005/1394/1399

AUTHOR: Chevy*chelov, A. D.

TITLE: Refinement of the Bueche criterion for breaking of a glassy polymer under applied load in light of the kinetic theory of strength

SOURCE: Fizika tverdogo tela, v. 5, no. 5, 1963, 1394-1399

TOPIC TAGS: glassy polymer, kinetic theory of strength, tensile strength, polymer chain, chemical bond, polyethylene, polyethylene tetrafluoride, polyamide, caprone, polypropylene, polystyrene, polyvinyl chloride

ABSTRACT: The author has solved the kinetic equations for breaking of oriented glassy polymers under applied load on the assumption that the load is uniformly distributed over all unbroken chains and that the life of the chemical bond is a function of the strains in polymer segments belonging to the rupturing bond. The results for time of breaking have been compared with the theory on strength and with the approximate criterion for breaking as proposed by Bueche. It has been shown that the qualitative course of time dependence of breaking on applied load, according to Bueche's criterion and according to the present theory, is the same for each, but the criterion of Bueche gives a somewhat lower time for breaking.

Card 1/82

L 13036-63

ACCESSION NR: AP3000619

10

It has been established that the ordinary theory of tensile strength is a restricted case of kinetic theory at zero temperature. In an approximation of noninteracting chains the author has computed the ultimate strength of a number of actual polymers (polyethylene, polyethylene tetrafluoride, polyamide (carbone), polypropylene, polystyrene, polyvinyl chloride), taking into account that the polymer chains form a simple zigzagging coil. The results obtained have been compared with experimental data from various sources, particularly from S. N. Zhurkov and S. A. Abasov (Vy'sokomol. soved., 3, 441, 450, 1961), and are in good agreement. The slope of the curve for breaking time proved to be less than experimental values by a factor of 7-10, but this is explained by the fact that the load is not uniformly distributed over all unbroken chains in actual polymers (since they are not fully oriented) but only over a small part of them. "I take this opportunity to express my deep thanks to S. N. Zhurkov, corresponding member of the AN SSSR, and to Professor A. I. Gubanov for their interest in the work and for useful discussions." Orig. art. has: 3 figures, 2 tables, and 15 formulas.

ASSOCIATION: Fiziko-tehnicheskiy institut im. A. F. Ioffe AN SSSR, Leningrad
(Physical and Technical Institute, Academy of Sciences, SSSR)

Card 2/2 2

GUBANOV, A.I.; CHEVYCHELOV, A.D.

Effect of intermolecular interaction on the strength of vitreous
polymers. Fiz. tver tela 5 no.9:2599-2608 S '63. (MIRA 16:10)

1. Fiziko-tekhnicheskiiy institut im. A.F.Ioffe AN SSSR, Leningrad.

CHEVYCHELOV, A.D.

Theory of activation barrier of the thermal breaking of
macromolecules in an amorphous-crystalline polymer.
Vysokom. soed. 8 no. 1:49-55 Ja '66 (MIRA 19:1)

1. Fiziko-tekhmicheskiy institut imeni Ioffe. Submitted
February 11, 1965.

POLAND

CHEW, Geoffrey F.

Lawrence Radiation Laboratory, Univ. of Calif.,
Berkeley, Calif.

Crakow, Postepy fizyki, No 6, Nov-Dec 1965, pp 647-56

"Problematic role of the time-space continuum in
subatomic physics."

POLAND

CHEW, Geoffrey F.

Lawrence Radiation Laboratory, Univ. of Calif.,
Berkeley, California

Crakow, Postępy fizyki, No 6, Nov-Dec 1965, pp 657-66

"Possibility of analysis as a basic principle of
physics." (Paper presented at a symposium on
modern natural philosophy in Pisa, 26 Jan 1965.)

L 61043-65

ACCESSION NR: AP5013912

the volume between the vortices. Extensive use is made of the analogy between the equations of motion of the superfluid, and the equations of the Ginzburg-Landau theory, used by A. A. Abrikosov (ZhETF v. 32, 1442, 1957), to explain the properties of superconductors of the second kind. This property of networks of quantum vortices is indicative of the principal difference between the wave function phases and the velocity potentials of the networks of geometrically identical classical and quantum vortices. This difference accounts for the capacity of the quantum vortices to create a rigid two-dimensional network. Orig. art. has: 14 formulas.

ASSOCIATION: Institut fiziki Akademii nauk Gruzinskoy SSR (Institute of Physics, Academy of Sciences, Georgian SSR)

SUBMITTED: 26Dec64

ENCL: 00

SUB CODE: ME, IC

NR REF SOV: 007

OTHER: 005

Card 2/2

MAMALADZE, Yu.G.; KHARADZE, G.A.; CHEYSHVILI, O.D.

Penetration of polarized neutrons through a superconductor in the mixed state. Zhur. eksp. i teor. fiz. 49 no.3:925-929 S '65.

1. Institut fiziki AN GruzSSR.

(MIRA 18:10)

CHEZHIN, V.A., inzh. [deceased]; LABETSKIY, K.I., inzh.

Construction of a reinforced concrete bridge in the city
of Volkhov. Transp. stroi. 15 no.11:10-12 N '65.
(MIRA 18:11)

CHERVYAKOVSKIY, N.Ya.; VOL'FSON, T.I.

Hyaluronidase activity in the blood and urine in cardiac edema.
Kardiologiya no.3:82 '65.

(MIRA 18:10)

1. Kafedra voyenno-morskoy i gospiatal'noy terapii (nachal'nik -
prof. Z.M.Volynskiy) Voyenno-meditsinskoy ordena Ienina akademii
imeni S.M.Kirova, Leningrad.

CHEYCHIS, I. [Ceicys, J.]

Channels of fertility. NTO 7 no.3:45-47 Mr '65.

(MIRA 18:5)

1. Predsedatel' byuro sektiis m-llioratsii Litovskogo respublikanskogo
pravleniya Nauchno-tekhnicheskogo obshchestva sel'skogo khozyaystva.

88204

16.3000
16.3800

S/020/60/134/002/029/041XX
C 111/ C 333

AUTHORS: Yevgrafov, M. A., Cheyis, I. A.

TITLE: Extension of Phragmen-Lindelöf's Theorem on Analytic Functions to Harmonic Functions in Space

PERIODICAL: Doklady Akademii nauk SSSR, 1960, Vol. 134, No. 2, pp. 259-262

TEXT: Theorem 1: Let $u(r, \varphi, x)$ be a harmonic function in the cylinder $r \leq a$, $0 \leq \varphi < 2\pi$, $-\infty < x < \infty$. If the conditions

$$(1) u(a, \varphi, x) = 0, \quad \left| \frac{\partial u}{\partial r}(a, \varphi, x) \right| < c$$

$$(2) \max_{(\varphi, x)} |u(r, \varphi, x)| < c \exp e^{\pi|x|/(2+\varepsilon)a}, \quad \varepsilon > 0$$

are satisfied, then $u(r, \varphi, x) \equiv 0$.

Theorem 2: Let $u(r, \theta, \varphi)$ be a harmonic function in the cone $0 < r < \infty$, $0 \leq \varphi < 2\pi$, $0 \leq \theta \leq \theta_0 < \pi$. If

$$(1') u(r, \theta_0, \varphi) = 0 \quad \left| \frac{\partial u}{\partial \theta}(r, \theta_0, \varphi) \right| < c$$

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$$(2') \max_{(\theta, \varphi)} |u(r, \theta, \varphi)| < c \exp \left(r + \frac{1}{r} \right)^{\pi/2\theta_0 - \varepsilon}, \quad \varepsilon > 0$$

are satisfied, then $u(r, \theta, \varphi) = 0$.

The proofs are based on: Theorem 3: Let $F(z) = \sum_{n=1}^{\infty} a_n e^{\lambda_n z}$
be an entire function and

$$(3) |a_n|^{1/n} < \frac{c}{n^{2+\varepsilon}}, \quad \varepsilon > 0$$

$$(4) \lim_{n \rightarrow \infty} \frac{n}{\lambda_n} = \alpha, \quad 0 < \alpha < \infty, \quad \lambda_n > 0$$

If there $|F(x)| < c, -\infty < x < \infty$, then $F(z) \equiv 0$. The proof
of theorem 3 is based on:

Lemma 1: If

$$(6) |F(t)| < c e^{-\delta|t|}, \quad -\infty < t < \infty, \quad 0 < \delta < \varepsilon,$$

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then $F_{\delta}(x + iy)$ is regular in

(7) $-\infty < x < \infty$, $|y| \leq \pi/2\delta - \eta$, $\eta > 0$

and satisfies there the inequality

$$(8) |F_{\delta}(x + iy)| < c e^{-\delta|x|}$$

Lemma 2: If $F(z) = \sum_{n=1}^{\infty} a_n e^{\lambda_n z}$ is an entire function, and if
(3), (4) are satisfied, while $\delta > 1/(2+\epsilon)\infty$, then

$$(9) F_{\delta}(z) = \sum_{n=1}^{\infty} a_n \Gamma\left(\frac{\lambda_n}{\delta} + 1\right) e^{\lambda_n z}$$

Lemma 3: Let $f(t + i\lambda)$ be regular in $|\lambda| \leq \gamma$, $-\infty < t < \infty$,
and assume that it satisfies there the inequality

$$|f(t + i\lambda)| < c e^{-\delta|t|} \quad . \text{ Then for the function}$$

$$\varphi(z) = \int_{-\infty}^{\infty} f(t) e^{-tz} dt \text{ regular in } |\operatorname{Re} z| < \delta \text{ it holds the}$$

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estimation $|\varphi(iy)| < c e^{-\gamma|y|}$

Lemma 4: Let denote

$$(10) \quad G_{\rho}(z) = \prod_{n=1}^{\infty} \left(1 - \frac{z^2}{\lambda_n^2}\right) \int_{-\infty}^{\infty} F_{\rho}(t) e^{-tz} dt, \quad \rho > \frac{1}{(2+\varepsilon)\alpha}.$$

The function $G_{\rho}(z)$ is analytically continuable into the semiplane
 $\operatorname{Re} z \geq 0$ and satisfies there the inequalities

$$(11) \quad |G_{\rho}(iy)| < c e^{-\pi|y|} \left(\alpha - \frac{1}{2\rho} + \varepsilon_{\rho}\right) \quad (\varepsilon_{\rho} > 0 \text{ arbitrary})$$

$$(12) \quad |G_{\rho}(z)| < c e^{b|z|}$$

S. N. Mergel'yan is mentioned in the paper. There are 4 references:
2 Soviet, 1 English and 1 American.

PRESENTED: May 3, 1960, by M. V. Keldysh, Academician

SUBMITTED: April 28, 1960

Card 4/4

L 21806-66 EWT(m)/EWP(t) IJP(c) JD
ACC NR: AP6012181

SOURCE CODE: UR/0386/66/003/008/0305/0309

AUTHOR: Kiknadze, L. V.; Mamaladze, Yu. G.; Chyashvili, O. D.

ORG: Institute of Physics, Academy of Sciences, Georgian SSR (Institut fiziki Akademii nauk Gruzinskoy SSR) 29 B

TITLE: State of liquid helium in the vicinity of the λ line

SOURCE: Zhurnal eksperimental'noy i teoreticheskoy fiziki. Pis'ma v redaktsiyu. Prilozheniye, v. 3, no. 8, 1966, 305-309

TOPIC TAGS: liquid helium, quantum liquid, superfluidity, critical point

ABSTRACT: The authors consider a vessel filled with liquid helium, such that at a certain depth the pressure corresponds to the λ line, and prove that contrary to expectations, the liquid helium should actually be either superfluid throughout or normal throughout. The proof is obtained by showing that the equation of the phenomenological superfluidity theory of Ginzburg and Pitayevskiy (ZhETF v. 34, 1240, 1958) admits of a nonzero solution, defined in the entire vessel. This means in turn that superfluidity is possible in the "normal" region, and under certain critical conditions the liquid remains normal even in the "superfluid" region. Since the rigorous proof entails certain mathematical difficulties, the authors

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merely confirm their conclusions by means of an approximate calculation, which is valid in the particular case of an "infinitely" deep "normal" region, covered by a "thin" superfluid layer. The critical thickness of the "superfluid" layer over the "infinitely" deep "normal" region is estimated at $\sim 2.2 \times 10^{-3}$ cm. The shift of the λ point, due to the external pressure, is estimated. Orig. art. has: 8 formulas.

SUB CODE: 20/ SUBM DATE: 31Jan66/ ORIG REF: 003/ OTH REF: 001

Cord 2/2

PB

L 22135-66 EWT(1)/EWP(m)/EWT(m)/EPF(n)-2/EWA(d)/EWP(t)/ETC(m)-6/EWA(1) IJP(c)
 ACC NR: AF6004935 JD/WW/03 SOURCE CODE: UR/0056/66/050/001/0169/0178
 AUTHOR: Mamaladze, Yu. G.; Cheyshvili, O. D.
 ORG: Institute of Physics, Academy of Sciences, Georgian SSR (Institut fiziki
 Akademii nauk Gruzinskoy SSR) 65
 TITLE: Flow of a ²¹superfluid liquid in porous media 64
 SOURCE: Zhurnal eksperimental'noy i teoreticheskoy fiziki, v. 50, no. 1, 1966, 169-178 B

TOPIC TAGS: superfluidity, wave function, quantum liquid, liquid helium, liquid flow, porosity, fluid density
 ABSTRACT: This is a continuation of earlier work (ZhETF Pis'ma v. 2, 123, 1965), where the authors proposed to describe the behavior of a superfluid situated in a porous medium by means of a phenomenological wave function averaged over a volume containing many pores. In the present article the authors derive a more general balance equation, and consider problems connected with the flow of the liquid, and especially the possibility of observing in helium II the analog of the Josephson direct current. To this end, a modified balance equation of phenomenological superfluidity theory is proposed, and it is shown that in the vicinity of a large helium II volume, the density of the superfluid component in a porous substance increases, the "penetration depth" of the effect increasing with decreasing pore

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ACC NR: AP6004935

size. The calculations are first made for the flow of the superfluid liquid in a single channel, assumed to be sufficiently narrow so that the normal component of the helium II remains immobile. This is followed by an extension of the results to include a large number of pores. It is shown that the density of the superfluid component of helium II bordering on a porous medium is increased by the propagation of the wave field of the condensate in the neighboring regions, and in this lies the analogy with the Josephson effect wherein current flows, without encountering resistance, through an insulator placed between two superconducting samples. The flow of the superfluid component through a porous medium through a porous partition immersed in a superfluid liquid is described and the critical fluxes at which the superfluidity breaks down are evaluated. The authors thank E. L. Andronikashvili for a remark made during the course of the work. Orig. art. has: 37 formulas.

SUB CODE: 20/ SUBM DATE: 12Jul65/ ORIG REF: 003/ OTH REF: 004

Card 2/2

BK

CZECHOSLOVAKIA/Cultivated Plants. Medicinal. Essential Oils. Poisons. M-9

Abs Jour: Ref Zhur-Biologiya, No 5, 1958, 20579.

Author : G. Chayka

Inst : Not given.

Title : The Eucalyptus and Its Possible Cultivation in Czechoslovakia
as a Pharmaceutical Raw Material.
(Evkalipt i vozmoszhnosti yego vyrashchivaniya v Chekho-
slovakii v kachestve farmatsevticheskogo syr'ya).

Orig Pub: Ziva, 1956, 4, No 5, 166-168.

Abstract: No abstract given.

Card : 1/1

CHEYKA, M.N.

~~XXXXXXXXXXXX~~

Homemade apparatus for geography. Geog.v shkole no.2:49-52 Mr-Apr '54.

(MLRA 7:1)

(Geography)

CHEYLYAKH, V.M., inzh.

Fields of application for two-stage system refrigerating machinery. Khol. tekhn. i tekhn. no.1:33-37 '65. (MIRA 18:9)

CHEYN, Ye.

"Chemical Constitution of Penicillin," Uspekhi Khimii, Vol. 18, p 23, 1949.

KAGAN, Z.S.; KRETOVICH, V.L.; CHEYSHNER, G.

Biosynthesis of isoleucine and its α , β -dihydroxy analog
in seedlings of various plants. Fiziol. rast. 10 no.4:458-
464 J1-Ag '63. (MIRA 16:8)

I. A.N. Bakh, Institute of Biochemistry, Academy of Sciences,
U.S.S.R. and the Technological Institute of Food Industry,
Moscow.

CHEYSHVILI, A.N.

New types of canned vegetables. Kons.i ov.prom. 18 no.2:16-18
F '63. (MIRA 16:2)

1. Kutaiskiy filial nauchno-issledovatel'skogo instituta
pishchevoy promyshlennosti soveta narodnogo khozyaystva
Gruzinskoy SSR.

(Vegetables, Canned)

CHEYSHVILI, A.P., inzh.

Knowledge of regulations is a guarantee of safe working conditions.
Bezop.truda v prom. 5 no.12:11-12 D '61. (MIRA 15:1)

1. Nachal'nik Yagodninskoy rayonnoy gornotekhnicheskoy inspeksii
Magadanskogo okruga Gosgortekhnadzora RSFSR.
(Blasting--Safety measures)

CHEYSHVILI, A.S.; MAMANTAVRISHVILI, D.G., prof., obshchestv.
red.

[Cybernetics in clinical medicine] Kibernetika v klinicheskoi meditsine. Tbilisi, Sabchota Sakartvelo, 1964. 89 p. (MIRA 18:7)

CHEYSHVILI, L.A.

Effect of castration on the structural features of the adrenals in
mammary cancer. Soob. AN Gruz. SSR 26 no. 3: 347-353 Mr '61.

(MIRA 14:4)

1. AN Gruzinskoy SSR, Institut eksperimental'noy i klinicheskoy
khirurgii i gematologii, Tbilisi, Predstavleno akademikom K.D.
Erostav.

(CASTRATION)

(ADRENAL GLANDS)

(MAMMARY GLANDS—CANCER)

TSINTSADZE, E.I.; DZINGOLO, A.A.; CHEYSHVILI, I.D.

Glycogen-forming function of the liver in rabbits with experimental hypercholesterinemia and atherosclerosis. Soob. AN Gruz. SSR 35 no.1:215-221 J1 '64. (MIRA 17:10)

1. Institut klinicheskoy i eksperimental'noy kardiologii imeni TSinandzgyrshvili. Predstavleno akademikom I.Ya. Tardishvili.

ERISTAVI, D.I.; BROUCHEK, F.I.; CHEYSHVILI, L.I.

Use of ion exchange resins for determining boron in natural
water. Report No. 1. Trudy GPI [Gruz.] no.5:3-16 '62.

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CHAYSHVILI, N. D.

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SO: U-2888, Letopis Zhurnal'nykh Statey, No. 1, 1949

MATINYAN, S.; CHEYSHVILI, O.

Polarization arising in the elastic scattering of fast deuterons
on protons and nucleon-nucleon interaction. Trudy Inst.fiz.AN
Grus.SSR 8:95-101 '62. (MIRA 16:2)
(Deuterons-Scattering) (Nuclear reactions)

~~CHAVCHANIDZE, V.V.~~

USSR/Nuclear Physics - Neutron distribution

FD-1855

Card 1/1 Pub. 145-15/25

Author : Chavchanidze, V. V., and Cheyshvili, O. D.

Title : Letter to the editor. ~~Finding the energy distribution function of neutrons~~
by the Markov method

Periodical : Zhur. eksp. i teor. fiz. 26, 369-371, March 1955

Abstract : The authors attempt to discuss certain problems in the theory of slowing of neutrons in the case where the slowing occurs as a result of elastic collisions with the nuclei of the moderator (A. I. Akhiezer and I. Ya. Pomeranchuk, Nekotoryye voprosy teorii yadra (Certain problems of nuclear theory), GITTL, 1950), namely by the use of the Markov method (V. V. Chavchanidze, ZhETF 26, 179, 185, 1954). Their results are also convenient for the case of thin layers of the moderator, where the number of elastic collisions is small (less than 25-30). Eight references; e.g. V. V. Chavchanidze, Dissertation, Tbilis State University, 1953).

Institution: Tbilis State University

Submitted : July 28, 1954

CHEYSHVILI, O.D.

SUBJECT USSR / PHYSICS
 AUTHOR CEJŠVILI, O.D.
 TITLE The Polarization of Deuterons in the Case of Elastic Scattering.
 PERIODICAL Žurn. eksp. i teor. fis, 30, 1147-1148 (1956)
 Publ. 6 / 1956 reviewed 9 / 1956

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PA - 1234

The present work investigates the scattering of polarized deuterons by nuclei with the spin zero. This problem is analogous to that of the triplet scattering of a nucleon by a nucleon. The polarization of the deuteron bundle is characterized by the 3 components of the spin vector and by the 5 components of the symmetrical two-step spin tensor with the trace zero. The deuteron is characterized before collision by a distorted plane wave:

$\psi_0 = \exp [i(\vec{k}_0 \vec{r} - \alpha \ln 2 Kr)] \chi_0$, where χ_0 is the initial spin function of the deuteron. The polarization of the incident deuteron bundle is characterized

by the quantities: $\vec{P}_{\text{beginning}} (\chi_0 \vec{S} \chi_0^+)$, $\langle T_{ik} \rangle_{\text{beg.}} = (\chi_0 T_{ik} \chi_0^+)$, where $T_{ik} = (1/2)(S_i S_k + S_k S_i) - (2/3)\delta_{ik}$ is the symmetrical tensor with the

trace zero. For the constant α it applies that $\alpha = Ze^2 \mu / K \hbar$. In order to ascertain the scattering amplitude the wave function of the system must be developed in a series according to the eigenfunctions of the operation $\vec{J}^2, J_z, \vec{L}, \vec{S}$. Here $\vec{J} = \vec{L} + \vec{S}$, \vec{L} is the orbital moment of the relative motion, \vec{S} - spin of the deuteron. Analysis with the help of the projective operators is explicitly written down.

CHEYSHVILLI, O.D.

4158

POLARIZATION OF ELASTICALLY SCATTERED DEUTERONS. Q.D. Chishevili (Leningrad Univ.). Soviet Phys JETP 9: 974-6418571 Jan.

Calculations are given concerning the elastic scattering of polarized deuterons by nuclei with spin $\frac{1}{2}$.

✓ CHEYSHVILI, O.D. Cand Phys-Math Sci -- (diss) "~~The~~ ^P polarization of
~~deutons~~ ^{during} ~~under~~ elastic dispersion." Tbilisi, 1957. 9 pp. (Tbilisi
State Univ im I.V. Stalin). 100 copies.
(KL, 8-58, 103)

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CHEYSHVILI, O.D.

USSR, Nuclear Physics - Nuclear Reaction

C-5

Abs Jour : Ref Zhur - Fizika, No 1, 1958, 518

Author : Cheyshvili, O.D.

Inst : Tbilisi University

Title : Deuteron Polarization in Elastic Scattering.

Orig Pub : Zh. eksperim. i teor. fiziki, 1957, 32, No 5, 1240-1242

Abstract : Using the Born approximation, the author calculates the polarization of the deuterons in elastic scattering by the nucleus, elastically scattered by a nucleus. For the interaction potential of the interaction between the deuteron and the nucleus, the author takes the sum $V_n(r_n) + V_p(r_p)$, averaged over the wave function of the internal motion of the deuteron. The value obtained for the intensity of the double scattering by carbon nuclei (with a deuteron energy of 167 Mev) is in rather good agreement with experimental data.

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On d+d Reactions

SOV/56-35-4-51/52

Here σ_t denotes the total cross section of the d+d reaction, W_i - the probability of the development of the d+d reaction through the i-th channel, σ_i - the cross section of the i-th reaction. Next, expressions are written down for the aforementioned 6 W_i . Finally, $\sigma_1/\sigma_3 \sim 5 \cdot 10^{-5}$ and $\sigma_2/\sigma_3 \sim 1.4 \cdot 10^{-2}$ is obtained. The ratio σ_4/σ_5 is probably nearly equal to 1. The reactions discussed here are, in the case of high energies, probably caused essentially by indirect interactions. There are 2 references, 1 of which is Soviet.

ASSOCIATION: Tbilisskiy gosudarstvennyy universitet
(Tbilisi State University)

SUBMITTED: April 25, 1958

Card 2/2

24(5), 24(3)

AUTHORS: Khutsishvili, G. R., Cheyshvili, O. D.

SOV/56-35-5-24/56

TITLE: Double Elastic Scattering of Deuterons in a Magnetic Field
(Dvoynoye uprugoye rasseyaniye deytronov v magnitnom pole)

PERIODICAL: Zhurnal eksperimental'noy i teoreticheskoy fiziki, 1958,
Vol 35, Nr 5, pp 1226-1231 (USSR)

ABSTRACT: Mendlowitz and Case (Mendlovits, Keyz) (Ref 1) carried out a theoretical investigation of the double elastic scattering of an electron beam for the case in which a magnetic field that is constant and homogeneous with respect to time acts upon it between the scatterers. Further, the authors of this paper theoretically investigated the double elastic scattering of a beam of particles with integral spin in the magnetic field. It is shown that in this case (contrary to what is the case with particles with half-integral spin) measurement of double elastic scattering in the magnetic field furnishes additional data concerning the scattering amplitudes. According to Cheyshvili (Ref 2) the following ansatz is used for the scattering amplitude (deuteron on nucleus with spin 0):

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SOV/56-35-5-24/56

Double Elastic Scattering of Deuterons in a Magnetic Field

$$F(\vartheta, \varphi) = A(\vartheta) + B(\vartheta)(\vec{S}\vec{n}) + C(\vartheta)(\vec{S}\vec{n})^2 + \frac{1}{2}D(\vartheta)\left\{(\vec{S}\vec{k}_0)(\vec{S}\vec{k}) + (\vec{S}\vec{k})(\vec{S}\vec{k}_0)\right\},$$

where \vec{k}_0 and \vec{k} denote the unit vectors in the direction of the deuteron momentum before and after scattering respectively, $n = [\vec{k}_0 \vec{k}] / \sin\vartheta$, the unit vector vertical to the scattering plane, S is the spin operator of the deuteron, and A, B, C, D are functions of the scattering angle and of deuteron energy. An expression is derived for the angular distribution of double elastic scattering, and the special cases for longitudinal and transversal fields are investigated. Formulae are also given for the differential cross section. There are 5 references, 2 of which are Soviet.

ASSOCIATION: Institut fiziki Akademii nauk Gruzinskoy SSR (Physics Institute of the Academy of Sciences of the Gruzinskaya SSR)

SUBMITTED: May 27, 1958

Card 2/2